

**GLOBAL OCEAN SURFACE WATER PARTIAL PRESSURE OF CO₂ DATABASE:
MEASUREMENTS PERFORMED DURING 1957–2018
(Version 2018)**

Contributed by
Taro Takahashi and Stewart C. Sutherland

Lamont-Doherty Earth Observatory, Columbia University, Palisades, NY 10964

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Prepared by
Alex Kozyr

National Centers for Environmental Information
National Oceanic and Atmospheric Administration
1315 East-West Hwy
Silver Spring, MD 20910
USA

CONTENTS

List of Figures	v
List of Tables	v
Abbreviations and Acronyms	vii
Abstract	ix
1. INTRODUCTION	1
2. OBJECTIVE	2
3. SUMMARY OF ADDITIONS AND CORRECTIONS IN ALL VERSIONS	2
4. METHODS OF COMPUTATION	5
5. QUALITY CONTROL	7
6. DATA SOURCES	8
7. DATA LISTING	10
8. HOW TO OBTAIN THE DATA AND DOCUMENTATION	11
9. ACKNOWLEDGEMENTS	12
10. REFERENCES	12

LIST OF FIGURES

1	Location of LDEO V2018 master database of sea surface pCO ₂ observations.	2
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LIST OF TABLES

1	List of data contributors to the global surface water pCO ₂ LDEO database	7
2	Data elements listed in the master data file “LDEO_Database_V2018.csv”	9
3	Data elements listed in the metadata file “LDEO_Database_Metadata_V2018.csv”	9

ABBREVIATIONS AND ACRONYMS

AOML	Atlantic Oceanographic and Meteorological Laboratory
CLIVAR	Climate Variability (Program)
CO ₂	carbon dioxide
CSIRO	Australian Commonwealth Scientific and Research Organization
DOE	U.S. Department of Energy
FTP	file transfer protocol
f CO ₂	fugacity of CO ₂
GEOSECS	Geochemical Ocean Sections Study
IGY	International Geophysical Year
JGOFS	Joint Global Ocean Flux Study
LDEO	Lamont-Doherty Earth Observatory
NCAR	National Center for Atmospheric Research
NCEI	National Centers for Environmental Information
NCEP	National Centers for Environmental Prediction
NDP	numeric data package
NOAA	National Oceanic and Atmospheric Administration
NSF	National Science Foundation
OCADS	Ocean Carbon Data System Project
ORNL	Oak Ridge National Laboratory
pCO ₂	partial pressure of CO ₂
PMEL	Pacific Marine Environmental Laboratory
SSS	sea surface salinity
SST	sea surface temperature
VOS	Volunteer Observing Ship
WOCE	World Ocean Circulation Experiment

ABSTRACT

Takahashi, T., S.C. Sutherland and A. Kozyr (2019). Global Ocean Surface Water Partial Pressure of CO₂ Database: Measurements Performed During 1957–2018 (LDEO Database Version 2018) (NCEI Accession 0160492). Version 3.3. NOAA National Centers for Environmental Information. Dataset.

Approximately 13.5 million measurements of surface water pCO₂ made over the global oceans during 1957-2018 have been processed to make a uniform data file in this Version 2018. Measurements made in open oceans as well as in coastal waters are included. The data assembled include only those measured using equilibrator-CO₂ analyzer systems, and have been quality-controlled based upon the stability of the system performance, the reliability of calibrations for CO₂ analysis and the internal consistency of data. We have added 531,520 data points comprised of 30 leg/cruise segments in this version. All of these were collected on the 5 ships in our current field program. These 5 ships operate primarily in high latitudes in both hemispheres and have built decades long records in these areas. R/V *Nathaniel B. Palmer*'s system has been operating since 1995, R/V *Laurence M. Gould*'s system since 2001, USCGC *Healy* since 2011, R/V *M. Langseth* since 2010, and R/V *Sikuliaq* since 2015. Our contribution to this database through many years of 3.19 million records is primarily for the polar and sub-polar seas. These underway data have been quality controlled and corrected for the time lag and temperature differences between the water intake and pCO₂ measurements. In order to allow re-examination of the data in the future, a number of measured parameters relevant to pCO₂ in seawater are listed. The overall uncertainty for the pCO₂ values listed is estimated to be ± 2.5 μ atm on the average. The names and institutional affiliations of the contributors are listed in Table 1. The documentations for the previous versions (V1.0, V2007, V2008, V2009, V2010, V2011, V2012, V2013, V2014, V2015, V2016, and V2017) of our database are available at NCEI via NCEI Ocean Carbon and Acidification Data Portal search: https://www.nodc.noaa.gov/oceanacidification/stewardship/data_portal.html.

The global pCO₂ data set is available free of charge as a numeric data package (NDP) from the OCADS: https://www.nodc.noaa.gov/ocads/oceans/LDEO_Underway_Database/. The NDP consists of the oceanographic data files and this printed documentation, which describes the procedures and methods used to obtain the data.

Keywords: carbon dioxide, partial pressure of CO₂, global carbon cycle, global ocean, equilibrator-CO₂ analyzer systems.

1. INTRODUCTION

Transfer of CO₂ from the atmosphere to the oceans is a critical process in the global carbon cycle, and is important for the future of the earth's climate. During the past decade, 6 to 10 Pg-C yr⁻¹ (1 Pg (peta grams) = 10¹⁵ grams = 1 Giga ton) has been emitted into the atmosphere from various anthropogenic sources including the combustion of fossil fuels, cement production, deforestation, land use changes and others. The annual uptake rate of CO₂ by the oceans has been estimated to be about 2 Pg-C yr⁻¹ for the past decade on the basis of various independent scientific methods, including changes in oxygen and CO₂ concentrations in the atmosphere (Keeling et al. 1996; Bender et al., 2005; Manning and Keeling, 2006), sea-air differences in CO₂ partial pressure (Takahashi et al., 2002; Takahashi et al., 2009), distribution of carbon isotopes (¹²C, ¹³C and ¹⁴C) (Gruber and Keeling, 2001; Quay et al, 2003), inversion of atmospheric CO₂ distribution data using atmospheric circulation models (Baker et al., 2006; Patra et al., 2006), and various global carbon cycle models (Sarmiento et al., 2000; Gruber and Sarmiento, 2002; Matsumoto et al., 2004; Gruber et al., 2009). Thus, about 25% of the anthropogenic CO₂ emissions are absorbed annually by the oceans, and about 50% remains in the atmosphere. As a result, the atmospheric CO₂ concentration is increasing at a mean rate of about 1.9 ppm yr⁻¹ (or 0.4% per year), and the concentration of CO₂ dissolved in surface ocean waters is also increasing, thus causing the acidification of ocean waters. Accurate documentation of changes that have occurred in the CO₂ chemistry of ocean waters is therefore important for understanding the fate of anthropogenic CO₂ released into the atmosphere as well as charting the future course of atmospheric CO₂ levels that would affect the earth's climate.

Carbon dioxide molecules react chemically with water to form bicarbonate (HCO₃⁻) and carbonate (CO₃⁼) ions, both of which do not communicate with the overlying air. Only about 0.5% of the total CO₂ molecules in seawater communicate with the air via gas exchange across the sea surface. This quantity is called the partial pressure of CO₂ (pCO₂), which represents the CO₂ vapor pressure. The seawater pCO₂ depends on the temperature, the total amount of CO₂ dissolved in seawater and the pH of seawater. Over the global ocean, it varies from about 100 µatm to 1000 µatm (1 µatm = 10⁻⁶ atm = 0.101325 Pascal). When seawater pCO₂ is smaller than the atmospheric pCO₂ (presently about 395 µatm), seawater takes up CO₂ from the overlying air; when it is greater than the atmospheric pCO₂, it emits CO₂ to the air. The rate of transfer of CO₂ across the sea surface is estimated by: (sea-air CO₂ flux) = (transfer coefficient) x (sea-air pCO₂ difference). The transfer coefficient depends primarily on the degree of turbulence near the interface, and is commonly expressed as a function of the square of wind speed. Since the time-space variation for atmospheric pCO₂ is much smaller than that for surface ocean water pCO₂, the magnitude of sea-air pCO₂ difference and hence the net flux is governed primarily by seawater pCO₂. Therefore, the interannual and seasonal variability for surface water pCO₂ is of particular interest.

The atmospheric CO₂ concentrations observed at a number of locations over the globe is summarized in GLOBALVIEW-CO2 (2012) and TRENDS ON LINE (2011). About 40 years ago, we started to assemble a global surface water pCO₂ data for time-space variability studies for the global oceans using the observations made by the Lamont-Doherty group. Many investigators from other institutions joined in and contributed their data to the database. The first publication on the global ocean pCO₂ and sea-air CO₂ flux (Takahashi et al., 1997) was based on about 0.25 million pCO₂ measurements; the second publication (Takahashi et al., 2002) about 1 million measurements. In the third publication in the Deep-Sea Research (Takahashi et al., 2009), a dataset that consisted of about 3 million pCO₂ measurements plus a number of other measured supporting parameters, has been analyzed and summarized. In the fourth publication in Marine Chemistry (2014), about 6 million pCO₂ measurements were used to produce the climatological maps of pCO₂, total CO₂ concentration, alkalinity, pH, and degree of saturation for CaCO₃ over the global ocean.

Recently, an international consortium named SOCAT has been established for assembling surface ocean water pCO₂ data produced by many international research groups (Bakker et al., 2016). The LDEO database serves as a major source for such the effort.

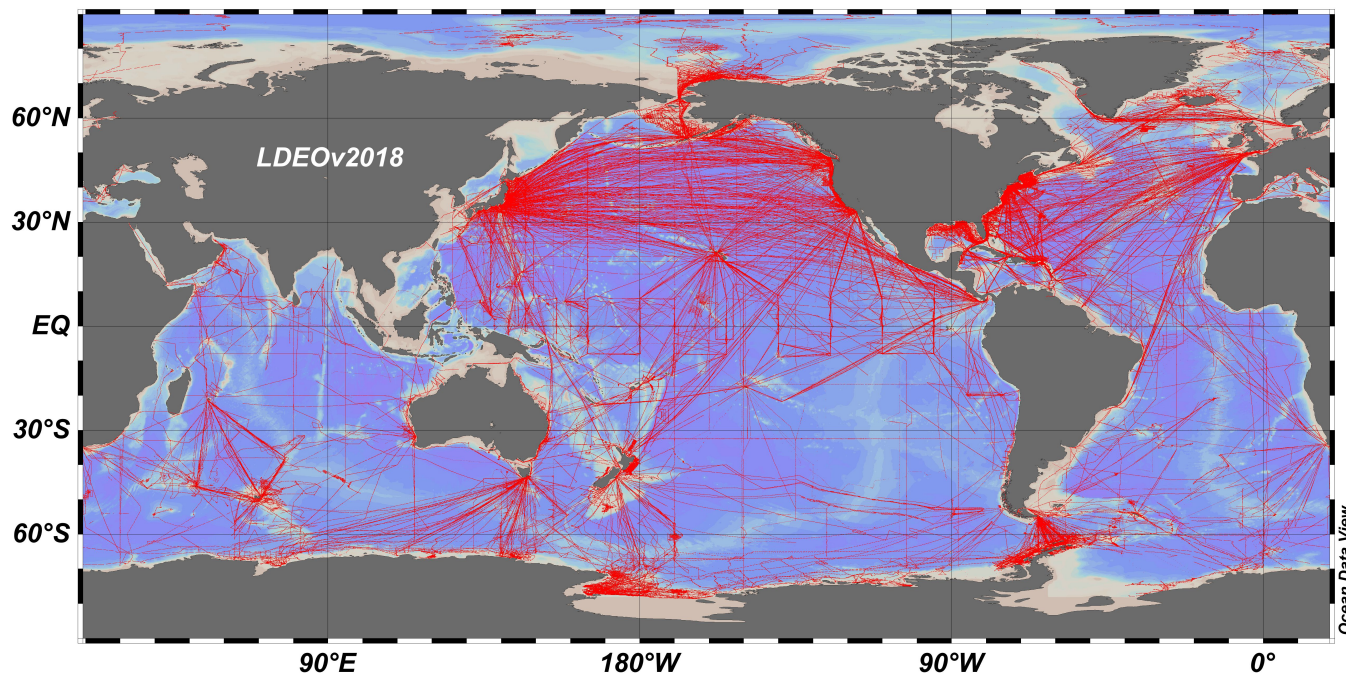


Fig. 1. Location of LDEO V2018 master database of sea surface pCO₂ observations.

2. OBJECTIVE

Modern investigation of ocean water pCO₂ started during the International Geophysical Year (IGY), 1957-1960, with the introduction of infrared CO₂ gas analyzer and gas-water equilibrator aboard oceanographic research vessels (Takahashi, 1961; Keeling et al., 1965; Keeling, 1965; Broecker and Takahashi, 1966; Keeling and Waterman, 1968; Keeling, 1968; Waterman et al., 2006-a, b, c and d). Combined with high-accuracy CO₂-air gas mixtures that were made available for the calibrations of the analyzers at sea, the quality of seawater pCO₂ data was greatly improved from that obtained in the pre-World War II era. During major oceanographic programs that took place after IGY, the CO₂ analyzers and equilibrators have been steadily improved to give more frequent observations and better accuracy, although the principles of the measurement remained essentially unchanged. Computers that are used for system control and data logging contributed significantly to the data quality. The deployment of GPS system improved the location information significantly. The objective of this data file is to assemble high-quality pCO₂ data obtained using the equilibrator-analyzer method, and to reprocess the data using a standardized method of computation, that will be described below. Therefore, the pCO₂ values listed in this database may differ from the original listings prepared by respective investigators.

3. SUMMARY OF ADDITIONS AND CORRECTIONS IN PREVIOUS VERSIONS

Summaries of the previous versions of the LDEO Database are reproduced in this section in order to show progression and changes made by the program.

VERSION 2017

Approximately 13.0 million measurements of surface water pCO₂ made over the global oceans during 1957-2017 have been processed to make a uniform data file in this Version 2017. Measurements made in open oceans as well as in coastal waters are included. The data assembled include only those measured using equilibrator-CO₂ analyzer systems, and have been quality-controlled based upon the stability of the system performance, the reliability of calibrations for CO₂ analysis and the internal consistency of data. We have added about 2.1 million data points comprised of 511 leg/cruise segments in this version. A total of 246,032 of these were collected on the 5 ships in our current field program. These 5 ships operate primarily in high latitudes in both hemispheres and have built decades long records in these areas. R/V *Nathaniel B. Palmer*'s system has been operating since 1995, R/V *Laurence M. Gould*'s system since 2001, USCGC *Healy* since 2011, R/V *M. Langseth* since 2010, and R/V *Sikuliaq* since 2015. Our contribution to this database through many years of 2.96 million records is primarily for the polar and sub-polar seas. These underway data have been quality controlled and corrected for the time lag and temperature differences between the water intake and pCO₂ measurements. The remainder of the 2.1 million new data were added from various web sites for Volunteer Observation Ships (VOS) operated by various agencies primarily NASA's Atlantic Oceanic and Marine Laboratory. In order to allow re-examination of the data in the future, a number of measured parameters relevant to pCO₂ in seawater are listed. The overall uncertainty for the pCO₂ values listed is estimated to be $\pm 2.5 \mu\text{atm}$ on the average. The names and institutional affiliations of the contributors are listed in Table 1. The documentations for the previous versions (V1.0, V2007, V2008, V2009, V2010, V2011, V2012, V2013, V2014, V2015, and 2016) of our database are available at NCEI via NCEI Ocean Carbon and Acidification Data Portal search:

VERSION 2016:

Approximately 10.8 million measurements of surface water pCO₂ made over the global oceans during 1957-2016 have been processed to make a uniform data file in this Version 2016. Measurements made in open oceans as well as in coastal waters are included. The data assembled include only those measured using equilibrator-CO₂ analyzer systems, and have been quality-controlled based upon the stability of the system performance, the reliability of calibrations for CO₂ analysis and the internal consistency of data. We have added about 300,000 data points comprised of 383 leg/cruise segments in this version. A total of 163,465 of these were collected on the 5 ships in our current field program, and the remainders were added from various web sites for Volunteer Observation Ships (VOS) operated by various agencies. These 5 ships operate primarily in high latitudes in both hemispheres and have built decades long records in these areas. R/V *Nathaniel B. Palmer*'s system has been operating since 1995, R/V *Laurence M. Gould*'s system since 2001, USCGC *Healy* since 2011, R/V *M. Langseth* since 2010, and R/V *Sikuliaq* since 2015. Our contribution to this database through many years exceeds 2.8 million records is primarily for the polar and sub-polar seas. The unique aspect of this version V2016 is the inclusion of an extensive re-analysis of our data from these 5 ships based on an updated algorithm for accounting for the delay between the time water passes the external Thermosalinograph (TSG) temperature probe and its arrival at our system. This varies from ship to ship depending on the geometry and speed of the uncontaminated seawater line. In order to allow re-examination of the data in the future, a number of measured parameters relevant to pCO₂ in seawater are listed. The overall uncertainty for the pCO₂ values listed is estimated to be $\pm 2.5 \mu\text{atm}$ on the average. The names and institutional affiliations of the contributors are listed in Table 1. The documentations for the previous versions (V1.0, V2007, V2008, V2009, V2010, V2011, V2012, V2013, V2014, and V2015) of our database are available at NCEI via NCEI Ocean Carbon and Acidification Data Portal search:

VERSION 2015:

A total of approximately 10.5 million measurements of surface water pCO₂ made over the global oceans during 1957-2015 have been processed to make a uniform data file in this Version 2015.

Measurements made in open oceans as well as in coastal waters are included. The data assembled include only those measured using equilibrator-CO₂ analyzer systems, and have been quality-controlled based upon the stability of the system performance, the reliability of calibrations for CO₂ analysis and the internal consistency of data. We have added about 1.4 million data points comprised of 461 leg/cruise segments in this version. A total of 200,000 of these were collected on the 5 ships in our current field program. The remainders were added from various web sites for Volunteer Observation Ships (VOS) operated by various agencies. Our ships operate primarily in high latitudes in both hemispheres and have built decades long records in these areas. R/V *Nathaniel B. Palmer*'s system has been operating since 1995, and R/V *Laurence M. Gould*'s system since 2001. Combined with the other three ships including USCGC *Healy*, R/V *Sikuliaq* and R/V *M. Langseth*, our contribution to this database through many years exceeds 2.3 million records primarily for the polar and sub-polar seas. These data have been corrected for the time-lag between the ship's intake port and the pCO₂ system as described in the technical report by Takahashi and Sutherland (2015).

VERSION 2014:

This version is referred to as Version 2014, and includes data collected through 31 December 2014. In this update, a total of about 100,000 data points comprising 467 leg/cruise segments have been added. This version includes the results of reanalysis for approximately 2.3 million pCO₂ measurements which were made in the surface waters of the polar oceans aboard RVIB Palmer, RV Gould and USCGC Healy using an underway method developed and operated by the Lamont-Doherty Earth Observatory (LDEO) group. Although these data were reported to the CDIAC in a series of previous reports after the preliminary quality control, a reanalysis of the data has shown that these reported data call for corrections that account for the transit time of pumped sample water from the intake port near the bow to the pCO₂ system located some distance away. The corrections applied to each data point are up to about ± 8 uatm for the Palmer and Gould data and ± 16 uatm for the Healy data, while the mean of the corrections for each cruise is less than ± 1 uatm.

VERSION 2013:

This version is referred to as Version 2013, and includes the data collected through 31 December 2013. In this update, a total of about 2,270,000 pCO₂ measurements made during 64 new cruise/ship files (including 170,770 new measurements made by the LDEO group) are added to the previous version 2012. A substantial number of files/data points from the participants of the NOAA sponsored Vessel of Opportunity (VOS) program were added. Several additional vessels as well as additional files from previously included vessels contributed significant number of the observations. When underway measurements were made across high SST gradient areas, errors up to about 5 μ atm may occur due to time lag of sample water transit from the intake to the equilibrator (Section 5). Because of the lack of the necessary information, the errors due to time lag are not corrected in this version.

VERSION 2012:

This version is referred to as Version 2012, and includes the data collected through 31 December 2012. In this update, about 277,000 pCO₂ measurements made during 29 new cruise/ship files are added to the previous version 2011. The new additions include the new data from our field operations aboard L.M. Gould, N.B. Palmer, Marcus G. Langseth, US Coast Guard Cutter Healy. These data files from Gould, Palmer, Langseth, and Healy are also posted on our LDEO website: <http://www.ldeo.columbia.edu/res/pi/CO2/>. In addition, several files from the Volunteer Observation Ships (VOS) program, M/V Explorer of the Seas, and 3 NOAA ships., which have recently been released for years 2010 – 2012, are combined into the master file of CDIAC <https://accession.nodc.noaa.gov/0059946/data/0-data/>

VERSION 2011:

This version is referred to as Version 2011 with approximately 6.4 million measurements of surface water partial pressure of CO₂, and includes the data collected through 31 December 2011. In this update, 209 new cruise/ship files are added to the previous version 2010. The new additions include the new data from our field operations aboard *L.M. Gould*, *N.B. Palmer*, *Marcus G. Langseth*, US Coast Guard Ship *Cutter Healy*. These individual data files are also posted on the CDIAC web site for Global Volunteer Observing Ship (VOS) Program: https://www.nodc.noaa.gov/ocads/oceans/VOS_Program/ and LDEO website: <http://www.ldeo.columbia.edu/res/pi/CO2/>.

VERSION 2010:

Approximately 5.2 million measurements of surface water partial pressure of CO₂ obtained over the global oceans during 1957–2010 are listed in the Lamont-Doherty Earth Observatory (LDEO) database, which includes open ocean and coastal water measurements. This version is referred to as Version 2010, and includes the data collected through 31 December 2010. In this update, 21 new cruise/ship files are added to the previous Version 2009 (see NDP-088_V2009). Other than routine updates from our Vessel of Opportunity Program such as *L.M. Gould*, *N.B. Palmer*, and *R/V Ronald H. Brown*, there are two new sources of data: *R/V Marcus G. Langseth*, operated by Lamont-Doherty Earth Observatory, and *S.A. Agulhas* operated by CSIR, South Africa.

VERSION 2009:

This version is referred to as Version 2009, and includes the data collected through 31 December 2009. In this update, 42 new cruise/ship files are added to the previous version 2008 (see NDP-088r) including several files for the IGY expeditions in 1957–1963. Four files were received from Dr. Ralph F. Keeling at Scripps Institute of Oceanography containing the data from the late Dr. Charles D. Keeling's work on the “1957 Downwind” and “1961 Monsoon”, “1962 Lusiad” and “1963 Lusiad” Expeditions, re-edited by Lee Waterman, one of the original investigators, to improve their accuracy (Waterman et al., 2006-a, -b, -c and -d. See references for URL). The other file is for the 1957–1958 *R/V Vema* Expedition by Takahashi (1961). Based on his independent calibrations, he reported a mean atmospheric CO₂ value (in dry air) of 315.3 ± 1.4 ppm (N = 28) over the Atlantic Sector of the Southern Ocean (51°S–57°S; 41°W–02°E) during March 4–24, 1958. C. D. Keeling reported a mean value of 314.1 ppm for March, 1958, at the South Pole Station, and 0.3 ± 0.1 ppm higher values at the Palmer and Amsterdam Island stations. These data sets are in agreement within their respective uncertainties, and therefore, no adjustment was made for the CO₂ values for gas samples equilibrated with seawater. All these data are reprocessed in a manner described below to maintain the consistency with the data in this database.

VERSION 2008:

This version is referred to as Version 2008, and includes the data collected through 31 December 2008. In this update, twenty-six new cruise/ship files are added to the previous version 2007. Dr. Nicolas Metzl of Universite Pierre et Marie Curie, Paris, kindly called our attention to discrepancies between his original and our data file listed in Version 2007. The discrepancies were caused by applying temperature correction to his data which were already corrected to SST. Affected are a total of 13,981 records for the southern Indian Ocean in the file names OISO for years 1998 and 2000 (File Name OISO), which were published in Metzl (2009). The errors range from $-29.6 \mu\text{atm}$ to $+1.3 \mu\text{atm}$ with an average of $-9.27 \pm 3.43 \mu\text{atm}$. In Version 2008, these errors are corrected, and a total of 67,403 new OISO data spanning years 2000–2008 are added.

VERSION 2007:

More than 4.1 million measurements of surface water partial pressure of CO₂ obtained over the global oceans during 1968–2007 are listed in the Lamont-Doherty Earth Observatory (LDEO) database, which includes open ocean and coastal water measurements. The data assembled include only those measured by equilibrator-CO₂ analyzer systems and have been quality-controlled based on the stability of the system performance, the reliability of calibrations for CO₂ analysis, and the internal consistency of

data. To allow re-examination of the data in the future, a number of measured parameters relevant to pCO₂ measurements are listed. The overall uncertainty for the pCO₂ values listed is estimated to be ± 2.5 µatm on the average.

VERSION 1.0 (2006):

More than 3 million measurements of surface water partial pressure of CO₂ obtained over the global oceans during 1968–2006 are listed in the Lamont-Doherty Earth Observatory (LDEO) database, which includes open ocean and coastal water measurements. The data assembled include only those measured by equilibrator-CO₂ analyzer systems and have been quality-controlled based on the stability of the system performance, the reliability of calibrations for CO₂ analysis, and the internal consistency of data. To allow re-examination of the data in the future, a number of measured parameters relevant to pCO₂ measurements are listed. The overall uncertainty for the pCO₂ values listed is estimated to be ± 2.5 µatm on the average.

4. METHODS OF COMPUTATION

The seawater pCO₂ data that are listed in this data file are based on direct measurements of seawater pCO₂ made using equilibrator-CO₂ analyzer systems. A large proportion of data listed is from semi-continuous underway pCO₂ systems with flow-through water, while many others are measurements for discrete water samples made at hydrographic stations along with measurements for other chemical and physical properties. Although different types of equilibrators (e. g. shower type, bubbling type, rotating disk type and membrane type in flow-through or closed circulation systems) and CO₂ gas analyzers (non-dispersive infrared analyzers and gas chromatographs of various designs) were employed, the results from different systems are accepted as long as analyzers were properly calibrated using validated CO₂-air gas mixtures and the carrier gas was equilibrated with seawater samples. Because of the diversity of methods used, it is not possible to present details of the method used by each research group who contributed data to this data file. Detailed methodology may be obtained directly from the original investigators who are listed in Table 1, or from the OCADS reports for specific expeditions.

It is important to point out that the methods used for computing CO₂ concentrations in equilibrated gas varied among groups. For example, some groups computed a least-squares fit of output readings for three or more standard gas mixtures to a quadratic equation and used it to calculate concentrations in samples. Other groups used four or five standard gas mixtures for calibrations, and fitted the data to a 4th order polynomial equation. Others used an output from linearization circuits of infrared analyzer (provided by the manufacturer), and linearly regressed three or more standard gas readings to obtain sample CO₂ concentrations. The outputs from a gas chromatograph are a linear function of CO₂ concentration, and hence a linear regression is used for calibration. These different data reduction methods yield CO₂ concentrations varying ± 1.5 ppm (or ± 1.5 µatm in pCO₂). However, we did not recompute the CO₂ values using a single uniform algorithm, and, instead, accepted CO₂ concentration values reported to us. Since different analyzers and different numbers of standard gas mixtures were used by respective groups, no single uniform data reduction scheme can be applied, and hence we relied on the judgment of each group for selecting the data reduction scheme most suited for their operational modes and skills. Measurements that were made using only one calibration gas mixture (not counting the CO₂-free air or nitrogen for establishing zero CO₂) are judged unreliable, and, hence, are not included in this database.

Using the reported CO₂ concentration values, the pCO₂ value in sample seawater at the equilibration temperature, (pCO₂)_{eq}, has been recomputed with the relationship:

$$(pCO_2)_{eq} = V_{co2} (P_{eq} - P_{water}),$$

where V_{CO_2} is the mole fraction concentration of CO_2 in dry carrier gas (V_{CO_2} is same as X_{CO_2} , which is often used in literature, and these quantities may be used interchangeably); P_{eq} is the total pressure of gas in the equilibrator; and P_{water} is the equilibrium water vapor pressure at temperature of equilibration, T_{eq} , and salinity. Since some equilibrators were operated open to the room air, P_{eq} values may be equal to the ship's interior pressure or to the barometric pressure outside the ship depending on the location of the equilibrator. When an equilibrator is located in an enclosed shipboard laboratory and is open to the room air, P_{eq} is the ambient pressure in the laboratory. While an equilibrator operated in an enclosed space, only the barometric pressure at sea surface was reported in some data sets, but not P_{eq} . In such cases, P_{eq} is assumed to be the reported barometric pressure at sea surface plus 3 mb, that represents an overpressure normally maintained inside a ship. This correction increases the $(pCO_2)_{sw}$ value by about 1 μatm . When the pressure was not reported, we used the climatological value in the nearest box from the National Centers for Environmental Prediction/National Center for Atmospheric Research (NCEP/NCAR) Reanalysis II Project file for the month of the observation.

The pCO_2 at in-situ seawater temperature is computed using an integrated form of the temperature effect for isochemical seawater, $(\partial \ln pCO_2 / \partial T)_{Sal, Alk, TCO_2}$ (Takahashi et al., 1993):

$$(pCO_2)_{sw} @ T_{in situ} = [(pCO_2)_{sw} @ T_{eq}] \text{Exp} \{ 0.0433 (T_{in situ} - T_{eq}) - 4.35 \times 10^{-5} [(T_{in situ})^2 - (T_{eq})^2] \}$$

where the “sw” and “eq” indicate the in situ and equilibrator conditions respectively. Throughout the computation, CO_2 gas is assumed to behave as an ideal gas that mixes with air and water vapor ideally. Although CO_2 fugacity is used in a number of published papers and data reports, we refrained from using the fugacity since it is computed differently from an investigator to another. Although we do not list the sea-air pCO_2 differences in this report, we recommend the formula below for the computation of atmospheric pCO_2 and the corresponding value for sea-air pCO_2 difference.

$$(pCO_2)_{air} = (V_{CO_2})_{air} (P_{baro} - P_{sw})$$

where P_{baro} is the barometric pressure at sea surface, and P_{sw} is the equilibrium water vapor pressure at the temperature and salinity for mixed layer water. The subscript “air” indicates the value for atmosphere samples.

The sea-air pCO_2 difference, ΔpCO_2 , is then computed using:

$$\Delta pCO_2 = (pCO_2)_{sw} - (pCO_2)_{air}$$

Since CO_2 is assumed to be an ideal gas for both $(pCO_2)_{sw}$ and $(pCO_2)_{air}$, the small effects of non-ideality should cancel due to differencing for ΔpCO_2 . Positive ΔpCO_2 values indicate that the sea is a source for atmospheric CO_2 , whereas negative values indicate that the sea is a sink.

Beginning with V2007 we added a column reporting the partial pressure of CO_2 in seawater in units of Pascals.

pCO₂ and fCO₂:

Values for the fugacity of CO_2 in seawater, fCO_2 , have been submitted to LDEO by some investigators. In principle, the chemical potential of CO_2 should be represented by fCO_2 . However, in practice, the fugacity values reported by various investigators are not clearly defined as to whether only the non-ideality arising from CO_2 - CO_2 molecular interactions is considered and/or that from CO_2 - H_2O interactions is also included. The original determination of the solubility of CO_2 in seawater by Murray and Riley (1971), which was conducted by equilibrating 1 atmosphere total pressure of pure CO_2 gas, appears to include implicitly the effects of CO_2 - H_2O interactions in gas phase above seawater samples, since the CO_2 gas above the sample water contained water vapor in equilibrium with the sample water at

respective temperatures. Weiss (1974) refined their data and presented a formulation for the CO₂ solubility under 1 atmosphere CO₂ fugacity. However, while Weiss's results account primarily for the CO₂-CO₂ interactions, it is probable that the effect of CO₂-H₂O interactions in the gaseous phase is also included. Subsequently, Weiss and Price (1980) proposed a formulation for the CO₂-H₂O interactions based upon Pitzer's correlations for interactions among gas molecules. Some researchers (e. g. Wanninkhof and Thoning, 1993) chose to correct Weiss's fugacity values using, in addition, the Weiss-Price CO₂-H₂O non-ideality effect. Whether such a procedure may constitute a double correction for the CO₂-H₂O interactions is not clear. Because of these ambiguities, we have chosen to report pCO₂ values assuming ideality. Since the mole fraction concentrations of CO₂ in equilibrated gas samples, temperature and pressure of equilibration are also included in the file, anyone who wishes to compute *f*CO₂ will be able to compute it with one's preferred algorithm for fugacity.

Since *f*CO₂ values are always smaller than the corresponding pCO₂ values by 1 to 2 µatm and the differences are large enough with respect to the precision of measurements and the mean global sea-air pCO₂ difference of about 10 µatm, they should not be used interchangeably with pCO₂. On the other hand, the sea-air difference in *f*CO₂ is numerically indistinguishable with measurement errors from the sea-air pCO₂ difference (pCO₂) as long as *f*CO₂ in air and pCO₂ in air are computed in the same manner as *f*CO₂ and pCO₂ for the seawater respectively.

In all OISO data files, only *f*CO₂ values are reported without pCO₂ values or CO₂ concentrations in dried equilibrated gas. We had to convert the reported *f*CO₂ values to pCO₂ values to make the OISO data consistent with the rest of the database. Since formulations used for *f*CO₂ calculation are not provided in their data report, it is not possible to re-compute pCO₂. Therefore, we use the following approximation for the conversion;

$$pCO_2 (\mu atm) = fCO_2 (\mu atm) \times [1.00436 - 4.669 \times 10^{-5} \times SST (^{\circ}C)]$$

This yields pCO₂ values greater than *f*CO₂ by 0.8 to 1.8 µatm (about 1.3 µatm on the average) in the range of temperatures and pCO₂ values encountered in the OISO study areas in the southern Indian Ocean.

5. QUALITY CONTROL

Determination of surface water pCO₂ requires measurements for a number of parameters in addition to the concentration of CO₂ in the carrier gas equilibrated with seawater. With the broad range of data sources and continued evolution of the measurement systems, it has not been possible to ensure that the observations contain all the necessary data elements with desired precisions. In some cases, pCO₂ or *f*CO₂ was reported without V_{CO₂}, sometimes with the pressure at the time of equilibration, sometimes not. Some files contained only the sea surface temperature and the pCO₂ at that temperature, but not the temperature and pCO₂ at equilibration. Ideally the incoming file should contain the mole fraction concentration of CO₂ in a parcel of dried air equilibrated with seawater (V_{CO₂}), the temperature at the time of equilibration and the equilibration pressure. When the pressure was not reported, we used the climatological value in the nearest box from the NCEP/NCAR Reanalysis II Project file for the month of the observation. We accepted data points that contained pCO₂ and sea surface temperature with or without the pressure.

Field data obtained in earlier years do not necessarily have all the required supporting measurements. For example, the temperatures of equilibration and water samples were recorded by hand at intervals of a few to several hours, and hence these values are often interpolated linearly with time between measurements. The salinity is measured only at hydrographic stations, and these values were interpolated between stations. When salinity was not measured during some expeditions, climatological values were assigned. Some data were unrecoverable and lost as a result of changes in computer systems (hard and software). For the past decade, computer controlled systems for underway surface water pCO₂

measurements have been deployed widely, and more complete high quality data sets have been obtained. In constructing this data file, we exercised our personal judgment for accepting and rejecting data, especially those collected in earlier years. For more recent data sets, water flow rates through the equilibrator and the temperatures of equilibrator water are recorded for each pCO₂ measurement. They often served as important criteria for identifying satisfactory operation of the equilibrator. Measurements made at reduced or stopped water flow conditions were rejected, as were those made at unusually rapid changes in the temperature of the water in an equilibrator. CO₂ gas analyzers are commonly calibrated using three or more gas mixtures of different CO₂ concentrations. Those measurements made using only a single calibration gas mixture are subject to unspecified uncertainties, and hence are rejected.

All data points have been examined individually as they were integrated into this database. While not perfect, we have tried to edit out obvious problems, erring on the side of leaving IN questionable ones. Considering differences in equilibrator designs, calibration methods and some interpolated parameters, we estimate that the uncertainty of the pCO₂ data presented in this data file is about ± 2.5 μatm on the average.

We have rejected the use of data quality flags in our files. The original team who performed the measurements is in the best position to accept or reject individual measurements. With this in mind, we either accept or reject data, using our best judgment, and erring on the side of including data points.

The LDEO underway pCO₂ systems consist of an air-water equilibrator and IR CO₂ analyzer. The pCO₂ values measured at equilibrator temperatures are recorded every 3 minutes along with the temperatures of equilibration (measured in equilibrator pool) and in-situ temperatures (measured by TSG). The equilibrator temperatures lag behind the in-situ temperatures due to the transit time of pumped seawater from the bow of ship to the pCO₂ system and to the residence time of water in the equilibrator pool. We have developed a simple statistical method for synchronizing TSG temperatures with equilibrator temperatures. The variance of the (Equilibrator temperature – TSG) temperature differences was minimized by advancing the equilibrator temperatures by one 3-minute time step at time. About 2.3 million pCO₂ data reported from the ships RVIB Palmer, RV Gould, RV Marcus G Langseth, USCGC Healy, and RV Sikuliaq have been reanalyzed for the time lag using the method. The time lag is estimated to be about 3 to 6 minutes for Palmer, Gould, Langseth and Sikuliaq, and up to 15 minutes for Healy. The time lag introduced errors in the uncorrected (equilibration – in-situ) temperatures of up to ± 0.5 °C for the Palmer and Gould, and up to ± 1 °C for the Healy. We find that the standard deviation of the (equilibrator - in-situ) temperature differences for the corrected data set is about ± 0.1 °C, which is consistent with the precision of temperature measurements. Accordingly, the seawater pCO₂ values, which were originally reported to CDIAC/OCADS by assuming no time lag, are corrected by up to ± 8 μatm ($= 0.5$ °C $\times 16$ $\mu\text{atm } ^\circ\text{C}^{-1}$ using $(\partial \ln \text{pCO}_2 / \partial T)$ of 0.0423 $^\circ\text{C}^{-1}$, Takahashi et al., 1993) for Palmer and Gould data and up to ± 16 μatm ($= 1.0$ °C $\times 16$ $\mu\text{atm } ^\circ\text{C}^{-1}$) for the Healy data. As a result of the corrections, the random errors in pCO₂ caused by the temperature-induced errors are reduced to within ± 1 μatm . The mean of uncorrected pCO₂ values for each cruise is found to be virtually unchanged from that of the corrected values. The details of the method used and examples are described in CDIAC/OCADS report by Takahashi and Sutherland (2015).

During some special expeditions (e.g. Hales and Takahashi, 2004), the time lag was determined precisely by using the arrival times of marker events (such as sharp temperature steps in water column), and the data were corrected for the measured time lag. Some of the pCO₂ data were obtained from bottled samples, in which the temperature and salinity were measured. Hence these data do not require time lag corrections. However, since most of the data listed in this database were obtained using underway systems, they may require time lag corrections unless the corrections were already applied by the data producers.

Seawater pCO₂ values that were computed using the alkalinity, total CO₂ concentration and pH data are not included in this data file because of their large uncertainties and potential biases resulting from different dissociation constants of carbonic, boric and other acids used for the computations.

During many cruises, atmospheric CO₂ concentrations were measured concurrently with surface water pCO₂. However, we suspect that many of them were contaminated by local sources (such as ship's

exhausts). Since atmospheric CO₂ concentrations vary from one air mass to another, especially in the northern hemisphere, distinguishing local contamination from natural variability is difficult. Hence, we decided to omit the atmospheric CO₂ data from this file. We recommend that the atmospheric CO₂ concentration data listed in the GLOBALVIEW-CO2 for computing the sea-air pCO₂ difference.

6. DATA SOURCES

The LDEO database Version 2018 lists 13.5 million surface ocean pCO₂ observations made since 1957. A large portion of the data is composed of observations from the continuous underway systems, such as those used aboard the National Science Foundation icebreakers *Nathaniel B. Palmer* and *Laurence B. Gould* operating primarily in the Southern Ocean, and *USCGC Healy* in the Arctic; and the research ships operated by the Atlantic Oceanographic and Meteorological Laboratory and the Pacific Marine Environmental Laboratory of the National Oceanic and Atmospheric Administration. The database also includes the observations made during a number of major national and international oceanographic programs such as Geochemical Ocean Sections Study (GEOSECS), Joint Global Ocean Flux Study (JGOFS), World Ocean Circulation Experiment (WOCE), Climate Variability (CLIVAR) Repeat Hydrography Project, Volunteer Observing Ships (VOS) Project and others, which were supported by the National Science Foundation, National Oceanic and Atmospheric Administration, and Department of Energy. A significant number of data have been contributed by international colleagues from many countries including Japan, Germany, France, UK, Iceland, Australia, Canada, the Netherlands, Norway, and others. Table 1 lists the major contributors for the database.

Table 1. List of data contributors to the global surface water pCO₂ LDEO database.

PI name [*]	Institution	Country
J. Akl	Integrated Marine Observing System (IMOS)	Australia
Thorarinn S. Arnarson	Marine Research Institute and University of Iceland	Iceland
Dorothee C. E. Bakker	School of Environmental Sciences, University of East Anglia	UK
Nicholas R. Bates	Bermuda Institute of Ocean Sciences	Bermuda
Richard Bellarby	Bjerknes Centre for Climate Research, University of Bergen	Norway
Wei-Jun Cai	School of Marine Science and Policy, University of Delaware	USA
Francisco Chavez	Monterey Bay Aquarium Research Institute	USA
Liqi Chen	Key Lab of Global Change and Marine Atmospheric Chemistry, Third Institute of Oceanography	China
David W. Chipman ^{**}	Lamont-Doherty Earth Observatory, Columbia University	USA
Cathy E. Cosca	Pacific Marine Environmental Laboratory, National Oceanographic and Atmospheric Administration	USA
Bruno Delille	Universite de Liege, Liege	Belgium
Hein J. W. de Baar	Netherland Institute for Sea Research	Netherlands
Richard A. Feely	Pacific Marine Environmental Laboratory, National Oceanographic and Atmospheric Administration	USA
Gernot Friederich	Monterey Bay Aquarium Research Institute	USA
John Goddard ^{**}	Lamont-Doherty Earth Observatory, Columbia University	USA
Burke Hales	College of Oceanic and Atmospheric Sciences, Oregon State University	USA
Mario Hoppema	Alfred Wegener Institute for Polar and Marine Research	Germany
Masao Ishii	Meteorological Research Institute	Japan
Truls Johannessen	Bjerknes Centre for Climate Research, University of Bergen	Norway
J. Joao	National University of Ireland, Galway	Ireland
V. Kitidis	Plymouth Marine Laboratory	UK

C. D. Keeling***	Scripps Institution of Oceanography, University of California San Diego	USA
Arne Körtzinger	Leibniz Institute of Marine Sciences	Germany
Nicolas Metzl	Laboratoire d'Océanographie et du Climat, Université Pierre et Marie Curie	France
A. Kuwata	National Research Institute of Fisheries Science	Japan
Takashi Midorikawa	Meteorological Research Institute	Japan
Ludger Mintrop	MARIANDA marine analytics and data	Germany
Pedro M. S. Monteiro	CSIR, South Africa, Jan Cilliers St, Stellenbosch 7599	South Africa
P. P. Murphy	Pacific Marine Environmental Laboratory, National Oceanographic and Atmospheric Administration	USA
David R. Munro	Department of Atmospheric and Oceanic Sciences and Institute of Arctic and Alpine Research Laboratory, University of Colorado, Boulder, CO	USA
A. Nakadate	Pollutants Chemical Analysis Center, Marine Division, Japan Meteorological Agency	Japan
Craig Neill	Integrated Marine Observing System (IMOS)	Australia
Timothy Newberger	Lamont-Doherty Earth Observatory, Columbia University	USA
Yukihiro Nojiri	Dept. of Earth & Environment Sciences, Hirosaki University and National Institute for Environmental Studies, Tsukuba	Japan
Jon Olafsson	Marine Research Institute and University of Iceland	Iceland
Are Olsen	Bjerknes Centre for Climate Research, University of Bergen	Norway
Tsueno Ono	National Research Institute of Fisheries Science	Japan
Christopher L. Sabine	Pacific Marine Environmental Laboratory, National Oceanographic and Atmospheric Administration	USA
S. Saito	Marine Division, Global Environment and Marine Department, Japan Meteorological Agency	Japan
Ute Schuster	School of Environmental Sciences, University of East Anglia, Norwich	UK
Tobias Steinhoff	Leibniz Institute of Marine Sciences	Germany
Stewart C. Sutherland	Lamont-Doherty Earth Observatory, Columbia University	USA
Peter Salomeh	Scripps Institution of Oceanography	USA
Colm Sweeney	Earth System Research Laboratory, National Oceanographic and Atmospheric Administration	USA
Taro Takahashi	Lamont-Doherty Earth Observatory, Columbia University	USA
Bronte Tilbrook	Australian Commonwealth Scientific and Research Organization (CSIRO) Marine and Atmospheric Research	Australia
B. Ward	National University of Ireland, Galway, IRELAND	Ireland
Rik Wanninkhof	Atlantic Oceanographic and Meteorological Laboratory, National Oceanographic and Atmospheric Administration	USA
Lee S. Waterman**	Scripps Institution of Oceanography, University of California San Diego	USA
Andrew Watson	School of Environmental Sciences, University of East Anglia	UK
Ray F. Weiss	Scripps Institution of Oceanography, University of California San Diego	USA
C. S. Wong	Institute of Ocean Sciences	Canada
H. Yoshikawa-Inoue	Graduate School of Environmental Earth Science, Hokkaido University	Japan

*The PI names are given in the alphabetical order.

**Retired
 ***Deceased

7. DATA LISTING

This NDP consists of two files: the file called “LDEO_Database_V2018.csv” contains all the numerical data, and the file “LDEO_Database_Metadata_V2018.csv” contains information on the data source, credit, institution, etc. The data elements in “LDEO_Database_V2018.csv” and their units are listed in Table 2; and those in “LDEO_Database_Metadata_V2018.csv” are explained in Table 3. Detailed cruise documentation (such as names of ship and ports and dates for departure and arrival) may be obtained directly from the respective PI’s.

Table 2. Data elements listed in the master data file “LDEO_Database_V2018.csv”

CRUISE_ID*	Internal LDEO Cruise Number
STN*	Station number assigned in the LDEO File
LAT	Latitude in decimal degrees (North is positive)
LON	Longitude in decimal degrees (East is positive)
MONTH/DAY/YEAR	Date
JDATE	Julian Date in decimal notation. Convention is 0001 UTC 1 Jan = 1.0
VCO2_SW**	Mole fraction concentration of CO ₂ (ppm) in dried air
TEMP_PCO2	Temperature at which pCO ₂ was measured in °C
SST	Sea Surface Temperature in °C
SSS	Sea Surface Salinity
PCO2_SST	Partial Pressure of CO ₂ in seawater (in units of microatmospheres) at the temperature in the SST column
PCO2_SSTPA	Partial Pressure of CO ₂ in seawater (in units of Pascals) at the temperature in the TEMP column
PCO2_TEQ	Partial Pressure of CO ₂ in seawater (in units of microatmospheres) at the temperature in theTEMP_PCO2 column. This is ordinarily the value that is actually measured
EQ_PBARO	Pressure in the equilibration vessel in units of millibars
SHIPPBARO	Barometric pressure in the outside air from the ship’s observation system in units of millibars

*Cross reference fields to the internal LDEO file name in case problems are discovered

**Missing data have the value -999.9

Table 3. Data elements listed in the metadata file “LDEO_Database_Metadata_V2018.csv”

CruiseID	Internal LDEO File Name
LEG	Leg Number. This is an arbitrary designation in many files because of our limit of four characters for a station number. For surface underway files that normally have many more than 9,999 observations we have broken the file into “legs.”
SHIP/EXPERIMENT	The name of the ship or other platform used, and experiment, project
CRUISE_NAME	Cruise name
OBSERVER	A notation of the person making the observation or responsible for running the collection system
DEPARTURE_PORT	Port of Departure
DEPARTURE_DATE	Date of Departure
ARRIV_PORT	Port of Arrival

ARRIV_DATE	Date of Arrival
COMMENT*	Air data source and general information

*Additional information may be available for a particular file if needed.

8. HOW TO OBTAIN THE DATA AND DOCUMENTATION

The global pCO₂ data set is available free of charge as a numeric data package (NDP) from the OCADS:https://www.nodc.noaa.gov/ocads/oceans/LDEO_Underway_Database/. The NDP consists of the oceanographic data files and this printed documentation, which describes the procedures and methods used to obtain the data.

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10. REFERENCES

- Baker D. F., R. M. Law, K. R. Gurney, P. Rayner, P. Peylin, A. S. Denning, P. Bousquet, L. Bruhwiler, Y.-H. Chen, P. Ciais, I. Y. Fung, M. Heimann, J. John, T. Maki, S. Maksyutov, K. Masari, M. Prather, B. Pak, S. Taguchi, and Z. Zhu. 2006. TransCom 3 inversion intercomparison: Impact of transport model errors on the interannual variability of regional CO₂ fluxes, 1988-2003. *Global Biogeochem. Cycles*, 20, GB1002, doi. 10.1029/2004GB2002439.
- Bakker, D.C.E., B. Pfeil, Camilla S. Landa, Nicolas Metzl, Kevin M.O'Brien, A.Olsen, Karl Smith, Cathy Cosca, Sumiko Harasawa, Stephen D. Jones, Shin-Ichiro Nakaoka, Yukihiro Nojiri, Ute Schuster, Tobias Steinhoff, ColmSweeney, Taro Takahashi, Bronte Tilbrook, Chisato Wada, Rik Wanninkhof, Simone R. Alin, Carlos F. Balestrini, Leticia Barbero, Nicholas R. Bates, Alejandro A. Bianchi, Frédéric Bonou, Jacqueline Boutin, Yann Bozec, Eugene F. Burger, Wei-Jun Cai, Robert D. Castle, Liqi Chen, Melissa Chierici, Kim Currie, Wiley Evans, Charles Featherstone, Richard A. Feely, Agneta Fransson, Catherine Goyet, Naomi Greenwood, Luke Gregor, Steven Hankin, Nick J. Hardman-Mountford, Jérôme Harlay, Judith Hauck, Mario Hoppema, Matthew P. Humphreys, Christopher W. Hunt, Betty Huss, J. Severino P. Ibáñez, Truls Johannessen, Ralph Keeling, Vassilis Kitidis, ArneKörtzinger, Alex Kozyr, Evangelia Krasakopoulou, Akira Kuwata, Peter Landschützer, Siv K. Lauvset, Nathalie Lefèvre, Claire LoMonaco, Ansley Manke, Jeremy T. Mathis, Liliane Merlivat, Frank J. Millero, Pedro M.S. Monteiro, David R. Munro, Akihiko Murata, Timothy Newberger, Abdirahman M. Omar, Tsuneo Ono, Kristina Paterson, David Pearce, Denis Pierrot, Lisa L. Robbins, Shu Saito, Joe Salisbury, Reiner Schlitzer, Bernd Schneider, Roland Schweitzer, Rainer Sieger, Ingunn Skjelvan, Kevin F. Sullivan, Stewart C. Sutherland,

- Adrienne J. Sutton, Kazuaki Tadokoro, Maciej Telszewski, Matthias Tuma, Steven M.A.C. vanHeuven, Doug Vandemark, Brian Ward, Andrew J. Watson, and Suqing Xu, (2016). A multi-decade record of high-quality fCO₂ data in version 3 of the Surface Ocean CO₂ Atlas (SOCAT). *Earth Syst. Sci. Data*, 8, 383-413, doi:10.5194/essd-8-383-2016.
- Bender, M.L., D.T. Ho, M.B. Hendricks, R. Mika, M.O. Battle, P.P. Tans T.J. Conway, B. Sturtevant, and N. Cassar. 2005. "Atmospheric O₂/N₂ changes, 1993–2002: Implications for the partitioning of fossil fuel CO₂ sequestration," *Global Biogeochemical Cycles* 19:GB4017, doi:10.1029/2004GB002410.
- Broecker, W.S. and T. Takahashi. 1966. Calcium carbonate precipitation on the Bahama Banks, *Journal of Geophysical Research* 71:1575–1602.
- Chavez, F. and T. Takahashi. 2007. Coastal oceans, Chapter 15, in "The First State of the Carbon Cycle Report (SOCCR): North American Carbon Budget and Implications for the Global Carbon Cycle". A report by the U.S. Climate Change Science Program and the Subcommittee on Global Change Research [King, A.W. L. Dilling. G.P. Zimmerman, D.M. Fairman, R.A. Houghton, G.H. Marland, A.Z. Rose, and T.J. Wilbanks (eds.)] National Ocean and Atmospheric Administration, Climate Program Office, Silver Spring, MD, USA, pp. 83–92.
- GLOBALVIEW-CO₂. *Cooperative Atmospheric Data Integration Project—Carbon Dioxide*. CD-ROM, NOAA CMDL, Boulder, Colorado (also available on Internet via anonymous FTP to ftp.cmdl.noaa.gov, Path: ccg/co2/GLOBALVIEW).
- Gruber, N., and C. D. Keeling. 2001. An improved estimate of the isotopic air-sea disequilibrium of CO₂: Implications for the oceanic uptake of anthropogenic CO₂. *Geophys. Res. Lett.*, 28(3), 555–558, doi:10.1029/2000GL011853.
- Gruber, N. and J.L. Sarmiento. 2002. "Large-scale biogeochemical-physical interactions in elemental cycles." In *The Sea*, Vol. 12:337–99. John Wiley & Sons, New York.
- Gruber, N., M. Gloor, S. E. Mikaloff Fletcher, S. C. Doney, S. Dutkiewicz, M. Follows, M. Gerber, A. R. Jacobson, F. Joos, K. Lindsay, D. Menemenlis, A. Mouchet, S. A. Müller, J. L. Sarmiento, and T. Takahashi. 2009. Oceanic sources, sinks, and transport of atmospheric CO₂. *Glob. Biogeochem. Cycles.*, 23, GB1005, doi:10.1029/2008GB003349.
- Hales, B. and Takahashi, T. (2004). High-resolution biogeochemical investigation of the Ross Sea, Antarctica, during the AESOPS (U. S. JGOFS) Program. *Global Biogeochem. Cycles*, 18, No. 3, GB3006, doi. 10.1029/2003GB002165.
- Keeling, C. D. 1965. Carbon dioxide in surface waters of the Pacific Ocean, 2, Calculation of the exchange with the atmosphere. *Journal of Geophysical Research*, 70:6099–102.
- Keeling, C. D., N. W. Rakestraw and L. S. Waterman. 1965. Carbon dioxide in surface waters of the Pacific Ocean, 1, Measurements of the distribution. *Journal of Geophysical Research*, 70:6087–98.
- Keeling, C. D. 1968. Carbon dioxide in surface waters, 4. Global distribution. *Journal of Geophysical Research*, 73:4529–4542.

- Keeling, C. D. and L.S. Waterman. 1968. Carbon dioxide in surface waters, 3. Measurements on Lusiad Expedition 1962-1963. *Journal of Geophysical Research*, 73:4529–42.
- Keeling, R.F., S.C. Piper, and M. Heinmann. 1996. “Global and hemispheric CO₂ sinks deduced from changes in atmospheric O₂ concentration,” *Nature* 381:218–21. Manning, A. C. and R. F. Keeling. 2006. Global oceanic and land biotic carbon sinks from the Scripps atmospheric oxygen flask sampling network. *Tellus B* 58:95–116.
- Majkut, J. D., J. L. Sarmiento, and K. B. Rodgers (2014), A growing oceanic carbon uptake: Results from an inversion study of surface *p*CO₂ data, *Global Biogeochem. Cycles*, 28, 335–351, doi:10.1002/2013GB004585.
- Manning, A. C. and Keeling, R. F. (2006). Global oceanic and land biotic carbon sinks from the Scripps atmospheric oxygen flask sampling network. *Tellus B*, 58, 95-116.
- Matsumoto, K. et al. 2004. Evaluation of ocean carbon cycle models with data-based metrics. *Geophys. Res. Lett.*, 31, LO7303, doi:10.1029/2003GL018970.
- Murray, C. N. and J. P. Riley. 1971. The solubility of gases in distilled water and seawater - IV. Carbon dioxide. *Deep-Sea Res.*, 18:533–541.
- Metzl, N. 2009. Decadal increase of oceanic carbon dioxide in the Southern Indian Ocean surface waters (1991–2007). *Deep-Sea Res. II*, doi:10.1016/j.dsr2.2008.12.007.
- Patra, P. K., Gurney, K. R., Denning, A. S., Maksyutov, S., Nakazawa, T., Baker, D., Bousquet, P., Bruhwiler, L., Chen, Y.-H., Ciais, P., Fan, S., Fung, I., Gloor, M., Heinmann, M., Higuchi, K., John, J., Law, R. M., Maki, T., Pak, B.C., Peylin, P., Prather, M., Rayner, P. J., Sarmiento, J., Tauchi, S., Takahashi, T. and Yuen, C.- W. 2006. Sensitivity of inverse estimation of annual mean CO₂ sources and sinks to ocean-only sites versus all-sites observational networks. *Geophys. Res. Lett.*, 33, LO5814, doi:10.1029/2005GL025403.
- Quay, P., R. Sommerup, T. Westby, J. Sutsman, and A. McNichol. 2003. “Changes in the ¹³C/¹²C of dissolved inorganic carbon in the ocean as a tracer of anthropogenic CO₂ uptake.” *Global Biogeochemical Cycles* 17, No. 1:DOI 10.1029/2001GB001817.
- Sarmiento, J. L., P. Monfray, E. Maier-Reimer, O. Aumont, R.J. Murnane, and J.C. Orr. 2000. “Sea-air CO₂ fluxes and carbon transport: A comparison of three ocean general circulation models.” *Global Biogeochemical Cycles* 14:1267–81.
- Sabine, C.L., R. A. Feely, N. Gruber, R. M. Key, K. Lee, J. L. Bullister, R. Wanninkhof, C. S. Wong, D. W. R. Wallace, B. Tilbrook, F. J. Millero, T.-H. Peng, A. Kozyr, T. Ono, A. F. Rios. 2004. The Oceanic Sink for Anthropogenic CO₂. *Science* 305(5682):367-371.
- Takahashi, T. 1961. “Carbon dioxide in the atmosphere and in the Atlantic Ocean water,” *Journal of Geophysical Research* 66:477–94.
- Takahashi, T., J. Olafsson, J. Goddard, D.W. Chipman, and S.C. Sutherland. 1993. “Seasonal variation of CO₂ and nutrients in the high-latitude surface oceans: A comparative study,” *Global Biogeochemical Cycles* 7:843–78.

- Takahashi, T., R.A. Feely, R. Weiss, R. Wanninkhof, D.W. Chipman, S.C. Sutherland, and T. Takahashi. 1997. "Global air-sea flux of CO₂: an estimate based on measurements of sea-air pCO₂ difference," *Proc. Nat. Acad. Sci.*, 94:8292–99.
- Takahashi, T., S.C. Sutherland, C. Sweeney, A. Poisson, N. Metzl, B. Tillbrook, N. Bates, R. Wanninkhof, R.A. Feely, C.L. Sabine, J. Olafsson, and Y. Nojiri. 2002. "Global sea-air CO₂ flux based on climatological surface ocean pCO₂, and seasonal biological and temperature effects," *Deep-Sea Research II*, 49:1601–22.
- Takahashi, T., S. C. Sutherland, R. Wanninkhof, C. Sweeney, R. A., Feely, D. W. Chipman, B. Hales, G. Friederich, F. Chavez, C. Sabine, A. Watson, D. C. E. Bakker, U. Schuster, N. Metzl, H. Yoshikawa-Inoue, M. Ishii, T. Midorikawa, Y. Nojiri, A. Kortzinger, T. Steinhoff, M. Hoppema, J. Olafsson, T. S. Arnarson, B. Tillbrook, T. Johannessen, A. Olsen, R. Bellerby, C. S. Wong, B. Delille, N. R. Bates, and H. J. W. de Baar. 2009. Climatological mean and decadal change in surface ocean pCO₂ and net sea-air CO₂ flux over the global oceans. *Deep-Sea Res. II*, 56, 554–577, doi:10.1016/j.dsr2.2008.12.009.
- Takahashi, T., Sutherland, S. C., Chipman, D. W., Goddard, J. G., Ho, C., Newberger, T., Sweeney, C., and Munro, D. R. (2014). Climatological Distributions of pH, pCO₂, Total CO₂, Alkalinity, and CaCO₃ Saturation in the Global Surface Ocean, and Temporal Changes at Selected Locations. *Marine Chemistry*, 164, 95–125. doi:10.1016/j.marchem.2014.06.004.
- Takahashi, T., S.C. Sutherland and A. Kozyr, (2015). Global Ocean Surface Water Partial Pressure of CO₂ Database: Measurements Performed During 1957–2014 (Version 2014). ORNL/CDIAC-161, NDP-088(V2014). Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, Tennessee, doi: 10.3334/CDIAC/OTG.NDP088(V2014).
- Takahashi, T. and S.C. Sutherland, (2015). Corrections applied to the LDEO underway pCO₂ measurements made aboard the R/V *Could*, R/V *Palmer* and USCGC *Healy*. CDIAC Report. Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, Tennessee, doi: 10.3334/CDIAC/OTG.LDEO_CORR.
- TRENDS on line. 2011. "A Compendium of Data on Global Change. Carbon Dioxide Information and Analysis Center," Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, TN <http://cdiac.ornl.gov/trends/co2/contents/html>).
- Wanninkhof, R. and K. Thoning. 1993. Measurement of fugacity of CO₂ in surface water using continuous and discrete sampling methods. *Marine Chem.*, 44:189–204.
- Waterman, L. S., P. P. Tans, T. Aten, C. D. Keeling, and T. A. Boden. 2006-a. Surface water and atmospheric underway carbon dioxide data obtained during Downwind (1957) expedition of Scripps Institution of Oceanography. Scripps Institution of Oceanography, Univ. California San Diego, La Jolla, CA http://scrippsco2.ucsd.edu/data/seawater_pco2/pco2_downwind_1957.csv
- Waterman, L. S., P. P. Tans, T. Aten, C. D. Keeling, and T. A. Boden. 2006-b. Surface water and atmospheric underway carbon dioxide data obtained during " Monsoon (1961) expedition of

Scripps Institution of Oceanography. Scripps Institution of Oceanography, Univ. California San Diego, La Jolla, CA

http://scrippsco2.ucsd.edu/data/seawater_pco2/pco2_monsoon_1961.csv

Waterman, L. S., P. P. Tans, T. Aten, C. D. Keeling, and T. A. Boden. 2006-c. Surface water and atmospheric underway carbon dioxide data obtained during Lusiad (1962) expedition of Scripps Institution of Oceanography. Scripps Institution of Oceanography, Univ. California San Diego, La Jolla, CA

http://scrippsco2.ucsd.edu/data/seawater_pco2/pco2_lusiad_1962.csv

Waterman, L. S., P. P. Tans, T. Aten, C. D. Keeling, and T. A. Boden. 2006-d. Surface water and atmospheric underway carbon dioxide data obtained during Lusiad (1963) expedition of Scripps Institution of Oceanography. Scripps Institution of Oceanography, Univ. California San Diego, La Jolla, CA

http://scrippsco2.ucsd.edu/data/seawater_pco2/pco2_lusiad_1963.csv

Weiss, R. F. 1974. Carbon dioxide in water and seawater: The solubility of a non-ideal gas. *Marine Chem.*, 2:203–215.

Weiss, R. F. and B. A. Price. 1980. Nitrous oxide solubility in water and seawater. *Marine Chem.*, 8:347–359.